

麦克默瑞提克(上海)仪器有限公司 Micromeritics Instrument (Shanghai) Ltd.

上海民生路 550 号 1505-1509 室, 邮编 200135 Suite 1505-1509, 550 Minsheng Rd., Shanghai, P.R. China 200135 www.micromeritics.com www.micromertics.com.cn t.sina.com.cn/micromeritics

Application Notes

使用 ASAP 2420 进行进行分子筛的微孔 分析

摘要

ASAP 2420 能够同时进行 6 个样品的微孔分析。因此,可通过此设备在 6 站同时进行分子筛类样品的微孔分析,以便进行数据比较。本文使用 Ar 作为吸附气,在 87K 的温度下进行分子筛材料的微孔测试。根据一般的测试经验, 实验用氮气作为吸附质进行经典的分子筛样品微孔分析,所需时间都较长, 有时甚至需要 5-7 天,但使用氩气作为吸附质进行分析,可将分析时间缩短到 2 天左右。

Application Note

Micropore Analysis of Zeolites Using the ASAP 2420

Introduction

Six zeolite samples were analyzed simultaneously using the ASAP 2420 equipped with the Micropore option. This option enables you to perform micropore analyses on all six ports at the same time, comparable to the High Throughput mode. These analyses were performed using argon as the adsorptive at 87 K. Typical micropore analyses using nitrogen may take as long as five to seven days. By using argon, the analyses can be shortened to as little as two days.

Materials

The following samples were used in the analyses:

Zeolite Type	Trade Name	SiO ₂ / Al ₂ O ₃
H-Y (FAU)	CBV600	5.2
H-Y (FAU)	CBV760	60
H-Y (FAU)	CBV901	80
H-β (BEA)	CP 811C-300	300
H-β (BEA)	CP 811E-75	75
ZSM-5 (MFI)	CBV3020	30

Preparation

First, the samples were prepared on the degas ports using the automated degas control. All samples were heated to 400 °C at 10 °C per minute and held at that temperature for two hours. Second, the samples were transferred to the analysis ports, where they were manually degassed at a temperature of 200 °C for one hour using lace-up heating mantles. Between the two degas procedures, each tube was weighed and the sample mass was calculated.

Analysis

The micropore analyses were performed with the lowpressure dosing option, and each sample tube was equipped with an isothermal jacket and seal frit.



initially filled to a level that was satisfactory as measured by the depth gauge. After the analyses were completed, the Dewars were refilled to a comparable level and a second series of singlepoint analyses was performed to determine the free space. A full isotherm was also obtained for each sample (see Figures 1 and 2).

Data

The linear and logarithm isotherm plots are shown in Figures 1 and 2. Results from the analyses show an interesting trend, which was expected to some degree. The H-Y samples all align very closely, as seen on the isotherm log plot. Interestingly, the H- β samples behaved somewhat differently from each other. The ZSM-5 also has a unique shape, which is to be expected considering it belongs to yet another class of zeolites. Shown in Figure 3 are the Horvath-Kawazoe differential plots giving the Saito-Foley cylinder pore volume.



One Micromeritics Drive, Norcross, Georgia 30093 T. (770) 662-3620 www.micromeritics.com



Figure 1. Isotherm with linear pressure axis. Samples are listed in the same order as in the table on page 1.





Figure 2. The same data as shown in Figure 1 with the pressure in log scale.

Figure 3. *The Horvath-Kawazoe pore volume distribution.*